

STUDENT NAME: _	 ()
HOMEROOM:		

Time Allowed: 60 Minutes

Class	Teacher	Dates	Periods
7 Respect	Morrison	Thursday 1 st Dec	1 & 2
7 Trust	Liu	Monday 28 th Nov	7 & 8
7 Joy	Morrison	Thursday 1st Dec	7 & 8
7 Care	Kitwood	Friday 2 nd Dec	5 & 6
7 Hope	Morrison	Wednesday 30 th Nov	4 & 5
7 Peace	Morrison	Thursday 1 st Dec	4 & 5

Unit title: What the world is made of

Key concept: Relationships **Related concepts:** Patterns

Global concept: Scientific and technical innovation

Statement of inquiry: We observe, look for patterns and relationships in order to help us understand the

natural world.

Inquiry question: What are the building blocks of our world?

Instructions:

- Topics included: Element symbols, Atomic structure, States of matter, Periodic table, Separation techniques, Group 1 elements
- You must write in blue or black ink only.
- Pencil should be used in drawing diagrams or graphs.
- Use of a dictionary, thesaurus is not allowed.
- Only this question paper, writing utensils and stationeries are permitted at the desk during the assessment.
- All other belongings must be placed under the desk.
- Read all of the paper carefully.
- You MUST use the following RUBRIC to ensure all Criteria are addressed.
- Periodic table is included

In this Summative Assessment you are assessed in Criteria A-Knowing and Understanding

*0- Does not meet any descriptors

Crit.	0	1-2	3-4	5-6	7-8
A					
	*	I tried to show my understanding on states of matter.	I was able to show my understanding on states of matter.	I demonstrated a good understanding on states of matter.	I demonstrated an excellent understanding on states of matter.
	*	I tried to determine the number of protons, electrons and neutrons in an atom.	I was able to determine the number of protons, electrons and neutrons in an atom with some errors.	I was able to determine the number of protons, electrons and neutrons in an atom mostly correct.	I was able to determine the number of protons, electrons and neutrons in an atom correctly .
		I tried to draw the electronic diagram of an atom.	I was able to draw the electronic diagram of an atom with some errors.	I was able to draw the electronic diagram of an atom mostly correct.	I was able to draw the electronic diagram of an atom correctly .
		I tried to distinguish among atom, element, compound and mixture.	I was able to distinguish among atom, element, compound and mixture with some errors.	I was able to distinguish among atom, element, compound and mixture mostly correct.	I was able to distinguish among atom, element, compound and mixture correctly.
		I tried to demonstrate my understanding of the periodic table with reference to reactivity in Group I.	I demonstrated a general understanding of the periodic table with reference to reactivity in Group I.	I demonstrated a good understanding of the periodic table with reference to reactivity in Group I.	I demonstrated an excellent understanding of the periodic table with a detailed reference to reactivity in Group I.
	*	I tried to draw a line graph.	I was able to draw a line graph with some errors.	I was able to draw a line graph with few errors.	I was able to draw a line graph successfully.
		I tried to describe separation techniques that I have learned.	I was able to describe separation techniques with some errors.	I was able to describe separation techniques with few errors.	I was able to describe separation techniques with key words and justification.